

AMENDMENTS TO THE CLAIMS:

This listing of claims will replace all prior versions, and listings, of claims in the application:

LISTING OF CLAIMS:

1. (Currently Amended) A process for the preparation of a hydrocyanation catalyst comprising an aqueous solution of at least one water-soluble phosphine and nickel, ~~values, which comprises said process comprising:~~ (a) admixing an aqueous solution of said at least one water-soluble phosphine with a nickel hydroxide to form a mixture, (b) adding hydrogen cyanide or a compound which generates hydrogen cyanide to the mixture thus formed, (c) stirring the resulting mixture until the nickel ~~values have~~ hydroxide is at least partially dissolved, and (d) reducing at least a portion of the nickel of said at least partially dissolved nickel ~~values having an oxidation state of greater than zero~~ hydroxide to the zero oxidation state.

2. (Currently Amended) The process as defined by Claim 1, comprising maintaining stirring in step (c) at a temperature of less than 100°C.

3. (Currently Amended) The process as defined by Claim 2, comprising maintaining stirring in step (c) at a temperature ranging from 20°C to 80°C.

4. (Currently Amended) The process as defined by Claim 1, comprising adding nickel in the zero oxidation state to the mixture of reaction ~~of reaction~~ [[,]] prior to the reduction stage (d) thereof.

5. (Currently Amended) The process as defined by Claim 1, ~~comprising (b) adding~~ wherein the hydrogen cyanide in step (b) is added in an amount at least equal to the stoichiometric amount for converting nickel hydroxide into nickel cyanide.

6. (Currently Amended) The process as defined by Claim 5, ~~comprising (b) adding a~~ wherein the amount of hydrogen cyanide added in step (b) is a 30% to 200% stoichiometric excess of the amount of hydrogen cyanide for converting nickel hydroxide into nickel cyanide.

7. (Currently Amended) The process as defined by Claim 1, wherein the amount of said at least one water-soluble phosphine, expressed as number of moles per 1 mol of nickel, ~~ranging~~ ranges from 0.5 and 2,000.

8. (Currently Amended) The process as defined by Claim 7, wherein said amount ~~ranging~~ ranges from 2 to 300.

9. (Currently Amended) The process as defined by Claim 1, ~~comprising (d)~~ wherein step

(d) comprises reducing said nickel ~~values~~ with gaseous hydrogen.

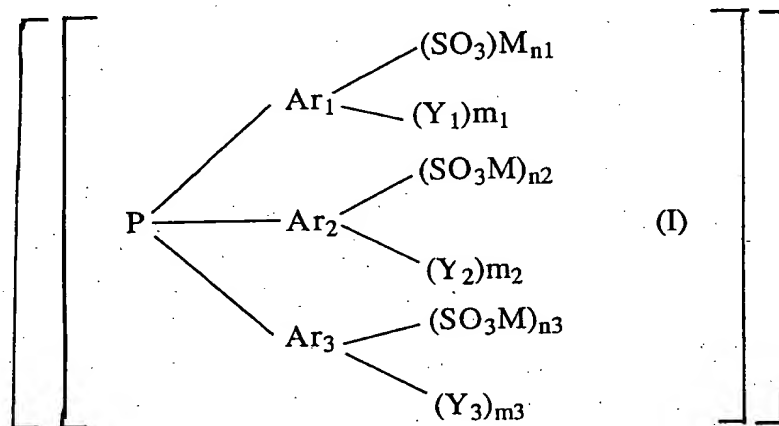
10. (Currently Amended) The process as defined by Claim 1, ~~comprising (d)~~ wherein step

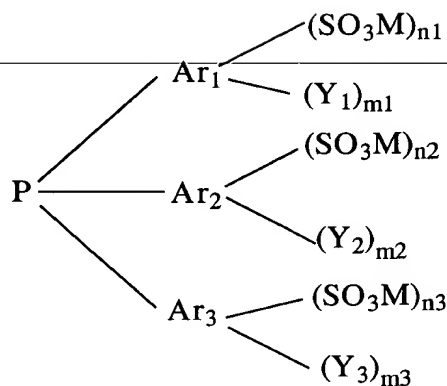
(d) comprises reducing said nickel ~~values~~ electrochemically.

11. (Currently Amended) The process as defined by Claim 1, ~~comprising (d)~~ wherein step

(d) comprises reducing said nickel ~~values~~ with an organic/inorganic reducing agent.

12. (Currently Amended) The process as defined by Claim 1, wherein said at least one water-soluble phosphine ~~having~~ has the structural formula (I):

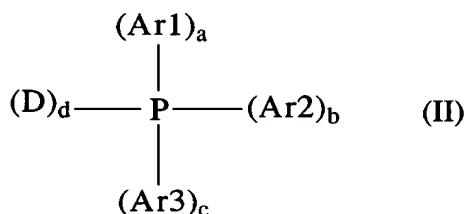




in which Ar_1 , Ar_2 and Ar_3 , which ~~may be~~ are identical or different, are each an aryl radical; Y_1 , Y_2 and Y_3 , which ~~may be~~ are identical or different, are each an alkyl radical having from 1 to 4 carbon atoms, an alkoxy radical having from 1 to 4 carbon atoms, a halogen atom, a CN group, an $[\text{NO}_2]$ NO_2 group, an OH group, an NR_1R_2 radical, wherein R_1 and R_2 , which ~~may be~~ are identical or different, are each an alkyl radical having from 1 to 4 carbon atoms; M is an inorganic or organic cationic residue selected, such that the compound of formula (I) is soluble in water, from the group consisting of H^+ , cations derived from alkali metals or alkaline earth metals, $\text{N}(\text{R}_3\text{R}_4\text{R}_5\text{R}_6)^+$, wherein R_3 , R_4 , R_5 and R_6 , which ~~may be~~ are identical or different, are each an alkyl radical having from 1 to 4 carbon atoms or a hydrogen atom, and ~~other~~ metal cations, the benzenesulfonic acid salts of which are soluble in water; m_1 , m_2 and m_3 which ~~may be~~ are identical or different, are each an integer ranging from 0 to 5; and n_1 , n_2 and n_3 , which ~~may be~~ are identical or different, are each an integer ranging from 0 to 3, at least one of these being equal to or greater than 1.

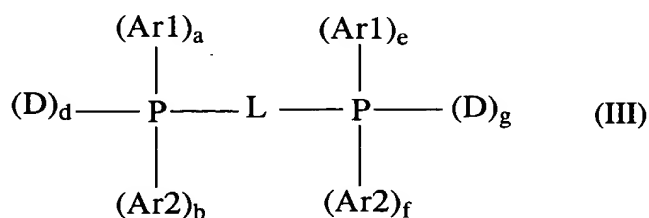
13. (Currently Amended) The process as defined by Claim 12, wherein said at least one water-soluble phosphine ~~having~~ has the structural formula (I) comprising an alkali metal or alkaline earth metal salt, ammonium salt, or quaternary ammonium salt of (3-sulfo-4-methylphenyl)di(4-methylphenyl)phosphine, (3-sulfo-4-methoxyphenyl)di(4-methoxyphenyl)phosphine, (3-sulfo-4-chlorophenyl)di(4-chlorophenyl)phosphine, di(3-sulfophenyl)phenylphosphine, di(4-sulfophenyl)phenylphosphine, di(3-sulfo-4-methylphenyl)(4-methylphenyl)phosphine, di(3-sulfo-4-methoxyphenyl)(4-methoxyphenyl)phosphine, di(3-sulfo-4-chlorophenyl)(4-chlorophenyl)phosphine, tri(3-sulfophenyl)phosphine, tri(4-sulfophenyl)phosphine, tri(3-sulfo-4-methylphenyl)phosphine, tri(3-sulfo-4-methoxyphenyl)phosphine, tri(3-sulfo-4-chlorophenyl)phosphine, (2-sulfo-4-methylphenyl)(3-sulfo-4-methylphenyl)(3,5-disulfo-4-methylphenyl)phosphine or (3-sulfophenyl)(3-sulfo-4-chlorophenyl)(3,5-disulfo-4-chlorophenyl)phosphine.

14. (Currently Amended) The process as defined by Claim 1, wherein said at least one water-soluble phosphine ~~having~~ has the structural formula (II) or (III):



in which Ar₁ and Ar₂, which ~~may be~~ are identical or different, are each aryl radicals or substituted ~~such~~ aryl radicals bearing one or more of the following substituents: alkyl or alkoxy radicals having from 1 to 4 carbon atoms, halogen atoms, the hydrophilic groups -COOM, -SO₃M or -PO₃M, wherein M is an inorganic or organic cationic residue selected from ~~among the group consisting of~~ hydrogen, cations derived from alkali metals or alkaline earth metals, ammonium cations $[[N(R)_4]]$ $-N(R)_4^+$ wherein the radicals R, which ~~may be~~ are identical or different, are each a hydrogen atom or an alkyl radical having from 1 to 4 carbon atoms, ~~and other~~ cations derived from metals, the arylcarboxylic acid, arylsulfonic acid or arylphosphonic acid salts of which are soluble in water, $[[N(R)_4]]$ ~~wherein the radicals R, which may be identical or different, are each a hydrogen atom or an alkyl radical having from 1 to 4 carbon atoms, or and~~ -OH; Ar₃ is a substituted aryl radical bearing one or more of the following substituents: alkyl or alkoxy radicals having from 1 to 4 carbon atoms, halogen atoms, the hydrophilic groups -COOM or -PO₃M, wherein M is an inorganic or organic cationic residue selected from ~~among the group consisting of~~ hydrogen, cations derived from alkali metals or alkaline earth metals, ammonium cations $[[N(R)_4]]$ $-N(R)_4^+$, wherein the radicals R, which ~~may be~~ are identical or different, are each a hydrogen atom or an alkyl radical having from 1 to 4 carbon atoms, ~~and other~~ metal cations, the arylcarboxylic acid or arylphosphonic acid salts of which are soluble in water, $[[N(R)_4]]$ ~~wherein the radicals R, which may be identical or different, are each a hydrogen atom or an alkyl radical having from 1 to 4 carbon atoms, or and~~ -OH, with the proviso that at least one of the substituents of Ar₃ is a hydrophilic group as defined

above; \underline{a} is 0 or 1; \underline{b} is 0 or 1; \underline{c} is an integer ranging from 0 to 3; D is an alkyl radical, a cycloalkyl radical or an alkyl or cycloalkyl radical substituted by one or more of the following substituents: an alkoxy radical having from 1 to 4 carbon atoms, a halogen atom, a hydrophilic group -COOM, -SO₃M or -PO₃M, wherein M is an inorganic or organic cationic residue selected from ~~among the group consisting of~~ hydrogen, cations derived from alkali metals or alkaline earth metals, ammonium cations $[-N(R)_4]$ $-N(R)_4^+$, wherein the radicals R, which ~~may be~~ are identical or different, are each a hydrogen atom or an alkyl radical having from 1 to 4 carbon atoms, and other metal cations, the arylcarboxylic acid, arylsulfonic acid or arylphosphonic acid salts of which are soluble in water, $[-N(R)_4]$ ~~wherein the radicals R, which may be identical or different, are each a hydrogen atom or an alkyl radical having from 1 to 4 carbon atoms~~ $[[,]]$ and -OH; \underline{d} is an integer ranging from 0 to 3; and the sum ($\underline{a} + \underline{b} + \underline{c} + \underline{d}$) is equal to 3 ~~or having the structural formula (II) $[[,]]$:~~



in which Ar1, Ar2 and D are as defined above for the formula (II); \underline{a} , \underline{b} , \underline{e} , and \underline{f} are each 0 or 1; \underline{d} and \underline{g} are each an integer ranging from 0 to 2; the sum ($\underline{a} + \underline{b} + \underline{d}$) is equal to 2; the sum ($\underline{e} + \underline{f} + \underline{g}$) is equal to 2; and L is a single ~~valency~~ bond between the two P atoms or a

divalent ~~hydrocarbonaceous~~ hydrocarbyl radical, or a radical ~~deriving~~ derived from a heterocycle comprising one or two oxygen, nitrogen or sulfur atoms in the ring, ~~these various cyclic radicals~~ said heterocyclic radical being bonded directly to one of the phosphorus atoms or both phosphorus atoms or being bonded to one of the phosphorus atoms or to both via a linear or branched alkylene radical having from 1 to 4 carbon atoms, with the proviso that the ring or rings which are optionally moieties of the divalent radical L optionally bear one or more substituents.

15. (Currently Amended) The process as defined by Claim 14, wherein said at least one water-soluble phosphine ~~having~~ has the structural formula (II) or (III) comprising tris(hydroxymethyl)phosphine, tris(2-hydroxyethyl)phosphine, tris(3-hydroxypropyl)phosphine, tris(2-carboxymethyl)phosphine, the sodium salt of tris(3-carboxyphenyl)phosphine, tris(3-carboxyethyl)phosphine, tris(4-trimethylammonioethyl)phosphine iodide, the sodium salt of tris(2-phosphonoethyl)phosphine, bis(2-carboxyethyl)phenylphosphine, the sodium salt of 2,2'-bis[di(sulfophenyl)phosphino]-1,1'-binaphthyl, the sodium salt of 1,2-bis[di(sulfophenyl)phosphinomethyl]cyclobutane (CBDTS), 1,2-bis(dihydroxymethylphosphino)ethane, 1,3-bis(dihydroxymethylphosphino)propane, or the sodium salt of 2,2'-bis[di(sulfophenyl)phosphinomethyl]-1,1'-binaphthyl.

16. (Currently Amended) ~~The~~ A hydrocyanation catalyst comprising an aqueous solution of at least one water-soluble phosphine and nickel ~~values~~, wherein said catalyst is prepared by the process which comprises (a) admixing an aqueous solution of said at least one water-soluble phosphine with a nickel hydroxide to form a mixture, (b) adding hydrogen cyanide or a compound which generates hydrogen cyanide to the mixture thus formed, (c) stirring the resulting mixture until the nickel ~~values have~~ hydroxide is at least partially dissolved, and (d) reducing at least a portion of the nickel of said at least partially dissolved nickel ~~values having an oxidation state of greater than zero~~ hydroxide to the zero oxidation state.